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Comparative molecular docking and toxicity between carbon-capped metal oxide nanoparticles and standard drugs in cancer and bacterial infections

Navid Mohammadjani¹⁰, Sahand Karimi¹, Musa Moetasam Zorab²⁰, Morahem Ashengroph¹⁺⁰, Mehran Alavi^{1,3+0}

¹Department of Biological Science, Faculty of Science, University of Kurdistan, Sanandaj, Kurdistan, Iran ²Department of Physics, University of Halabja, Kurdistan Region, Iraq

³Nanobiotechnology Department, Faculty of Innovative Science and Technology, Razi University, Kermanshah, Iran



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Abstract

Introduction: Nanoparticles (NPs) are of great interest in the design of various drugs due to their high surface-to-volume ratio, which result from their unique physicochemical properties. Because of the importance of examining the interactions between newly designed particles with different targets in the case of various diseases, techniques for examining the interactions between these particles with different targets, many of which are proteins, are now very common.

Methods: In this study, the interactions between metal oxide nanoparticles (MONPs) covered with a carbon layer (Ag₂O₃, CdO, CuO, Fe₂O₂, FeO, MgO, MnO, and ZnO NPs) and standard drugs related to the targets of Cancer and bacterial infections were investigated using the molecular docking technique with AutoDock 4.2.6 software tool. Finally, the PRO TOX-II online tool was used to compare the toxicity (LD₅₀) and molecular weight of these MONPs to standard drugs.



Results: According to the data obtained from the semi flexible molecular docking process, MgO and Fe₂O₂ NPs performed better than standard drugs in several cases. MONPs typically have a lower 50% lethal dose (LD_{50}) and a higher molecular weight than standard drugs. MONPs have shown a minor difference in binding energy for different targets in three diseases, which probably can be attributed to the specific physicochemical and pharmacophoric properties of MONPs. *Conclusion:* The toxicity of MONPs is one of the major challenges in the development of drugs based on them. According to the results of these molecular docking studies, MgO and Fe₂O₂ NPs had the highest efficiency among the investigated MONPs.

Introduction

Nowadays, one of the most important techniques in computer-aided drug design is molecular docking, which is used in drug design projects to investigate and measure the interactions between target proteins and drugs, synthetic or natural compounds with potentially therapeutic properties. AutoDock 4.2.6 is a critically important and widely utilized software in the field of molecular docking for various research studies,¹⁻³ which is a potential capability in molecular docking due to the ability to define the parameters of metal atoms that cannot be identified by much other software in the field of molecular docking.⁴ Nanotechnology has provided new options to treat various diseases in recent decades; however, many applications of this science, particularly in the design of drugs and medical diagnostic systems, have yet to be realized at the clinical level.5 The lack of biocompatibility and proper stability and toxicity of nanoparticles (NPs) produced by chemical and physical techniques, as well as the inappropriate scalability of NPs, are among the reasons for this. Fortunately, nanobiotechnology is helping to solve many of these issues.67 In fact, microorganisms and other living organisms, including animals and plants, have a unique ability in the biosynthesis of NPs with caps, which



*Corresponding authors: Morahem Ashengroph, Email: m.ashengroph@uok.ac.ir; Mehran Alavi, Email: mehranbio83@gmail.com

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significantly decreases toxicity, increases biocompatibility, and stability.^{8,9} The binding of nanomaterials to various proteins can cause physiological and pathological changes, such as the tendency of proteins to bind or the activation or inhibition.^{10,11} Silver oxide NPs are increasingly being used in food packaging, medical equipment, and food supplements because of their excellent antimicrobial properties.^{12,13} Iron oxide NPs have also been approved by food and drug regulatory organizations around the world for anemia treatment, and some studies have reported their antiviral ability.14 Furthermore, copper oxide NPs also have antimicrobial properties due to their high surface-to-volume ratio as well as specific morphological properties.15 Other NPs also have shown their potential ability in the production of reactive oxygen species (ROS) and the destruction of cancer cells, microbials cells, etc. ZnO NPs have a number of benefits, including lower toxicity and simple absorption by humans. ZnO NPs are regarded by the FDA as generally recognized as safe (GRAS) when compared to other MONPs.¹⁶ Multidrugresistant bacterial (MDR) infections, as one of the emerging challenges, have caused the development of new antibiotics and the use of innovative approaches in antibacterial drug design.¹⁷⁻¹⁹ The dihydropteroate synthase enzyme plays an important role in the synthesis of folic acid in Escherichia *coli*.²⁰ Penicillin-binding protein, in particular its 2X type, is a protein responsible for beta-lactam antibiotic group resistance, and it is a major contributor to the development of antibiotic resistance in a variety of bacteria against beta-lactam-based antibiotics.²¹ Aquaporin Z is another antibacterial target that functions in water exchange in the bacterial cell membrane.²² Bacterial DNA gyrase plays a key role in the translation and transcription of bacterial DNA because of its role in breaking two strands of DNA by the process of catalysis of the negative supercoil; therefore, it is one of the major targets of antibacterial pharmaceutical inhibitors.23-25 Filamenting temperaturesensitive mutant Z, or FtsZ, has been introduced as one of the new antibacterial therapeutic targets for the development of new antibiotics because of its important role in bacterial cell division and also conserved across different bacterial species. One of the primary benefits of targeting this protein is that it is absent in more advanced organisms, including humans, implying that antibiotics targeting this protein do not harm human cells.26 The DNA molecule has been a primary and traditional target for the development of anti-cancer drugs. these drugs can play a significant role in disabling cancer cells by creating bonds with the DNA structure and being placed inside the structure.²⁷ The hypoxia-inducible factor I, or HIF-1, is one of the transcription factors whose increase in the tumor microenvironment in response to hypoxia. This protein is one of the therapeutic targets in the treatment of certain types of cancer, such as liver cancer.²⁸ Several studies have shown that human leukemia inhibitory factor, or LIF, plays a key role in the development of pancreatic ductal adenocarcinoma tumors.²⁹ Caspases, such as caspase-3 and caspase-9, are cysteine proteases that play a key role in cancer cell apoptosis; therefore, the regulators of these proteins can be significant in antitumor functions.^{30, 31} Serum and glucocorticoid-regulated kinase 1 (SGK1), also known as one of the SGK isoforms and a member of the AGC family, is one of the therapeutic targets because it has a favorable impact on tumor cell survival, increases tumor invasiveness and adhesion, and stimulates tumor growth.³² Cell division protein kinase 2 (CDK2) is another type of protein whose overexpression, promotes cancer cell growth.33 The X-linked inhibitor of apoptosis protein (XIAP) is one of the proteins whose increased expression disrupts the cell apoptosis system and, as a result, the development of cancer; thus, protein inhibitors can play a role in cancer treatment via inhibition this protein.34 Telomerase overactivity has been observed in a variety of human cancers, suggesting that telomerase reverse transcriptase inhibitors could be used as antitumor therapeutic targets.35 Based on CRISPR dropout screens performed by Gao et al., the 3-hydroxy-3-methylglutaryl-coenzyme A enzyme or (HMG-CoA) reductase has been introduced as a therapeutic target for the antitumor activity of colon cancer stem cells, and inhibitors of this enzyme may have an anti-tumor role in this field.³⁶

In this study, the molecular docking technique was used to investigate the way introduced drugs interact with antibacterial and anticancer drug targets compared to MONPs with a conceptual carbon cap. Considering the need to develop new drugs for various diseases, as well as several studies that have pointed to the medicinal role of MONPs in the diseases mentioned above, this study used the semi flexible molecular docking process to obtain the conceptual and theoretical feasibility of the possible medicinal role of these MONPs. This study used MONPs capped with carbon layer to simulate MONPs that biosynthesized with different organisms. Because such caps are very diverse and it is not possible to introduce a specific cap as a representative of all such caps, only a carbon cap layer was used in this study. To further investigate the performance of MONPs, the toxicity of these particles was compared to standard drugs.

Materials and Methods Preparation of required files

The RCSB Protein Data Bank (https://www.rcsb.org/) was used to obtain the crystal structure files for the targets in this article. Antibacterial targets included 1AJ0, 1QME, 1RC2, 4URO, and 5V68, while 1BNA, 1EMR, 1H2N, 1RE1, 2AR9, 3HDN, 3IG7, 5CQG, 5OQW, 8DJM were anticancer targets.

All files downloaded from the RCSB PDB were in PDB format. The Drug Bank database (https://go.drugbank. com/) was then used to identify the main drugs of the targets. In this manner, the search in Drug Bank was performed based on the type and title of the desired target,

as well as the type of organism, and then the drug files in PDB format were saved. After storing all of the target files and standard drugs, it was time to prepare and design the MONPs. At this stage, 8 metal oxide nanostructures with dimensions $X \times Y \times Z = 5 \times 5 \times 5$ angstroms were produced by Materials Studio 2020 software, including Ag₂O₃, CdO, CuO, Fe₂O₃, FeO, MgO, MnO, and ZnO with a carbon layer cap. After the nanostructure was saved. It should be noted that the Open Babel GUI 3.1.1 software is used in the process of performing the steps if the file format needs to be changed.

Preparation of targets and prediction of their binding sites

The UCSF Chimera 1.16 software was used to remove extra atoms (such as water, ions, etc) from the PDB files, as well as chains and other extra parts from the target files, before saving the file in mol2 format. The process of adding hydrogen to the target structure and then calculating and adding the Gasteiger charge was then completed using the software AutoDock tools 1.5.7 (ADT). The targets were saved in the pdbqt format file at this point. The Discovery Studio 2021 software was used at this point to predict the possible binding sites of all targets, and the XYZ coordinates of the best ones (number one) according to natural ligand-protein interactions recorded in PDB, were chose for the molecular docking process.

Carbon capped-MONPs and drugs

UCSF Chimera 1.16 was used to add hydrogen and then Gasteiger charge added to the received drugs from DrugBank, after which the file was saved in mol2 format and read by ADT 1.5.7 software, then Torsion Tree was calculated with this software, and the bonds that could be rotated were confirmed.

Molecular docking process

The docking software that used in this study was AutoDock 4.2.6.⁴ The type of docking is semi flexible in this study.⁴ The GPF (Grid parameter file) and DPF (docking parameter file) were created in the first stage of the docking process using a genetic algorithm with the ADT 1.5.7. The grid box size was $120 \times 120 \times 120$ angstroms. Also, in the case of NPs, parameter and bond files was accompanied by a few changes, and the parameter of metal atoms used in the study were added. Before producing the GPF files and DPF, the new parameter and bond files, which contained the parameter information of the new metal atoms, were defined for the ADT software in each case. At this point, all of the files required to carry out the molecular docking procedure were ready.

Analysis of molecular docking results

The results from AutoDock 4.2.6 were analyzed by ADT 1.5.7 and Discovery Studio 2021 software. To evaluate the

results of drug docking with their targets, ADT software first saved the final virtual screening pdbqt file of the best binding pose run, and then the interactions between drugs and key residues and their types were analysed by ADT and Discovery Studio software using 3D and 2D visualizations. In addition, to validate the nanoparticle docking results with targets, ADT software and discovery studio checked the key residues involved in the NP-target interaction, and then Discovery Studio and ADT software evaluated the types of interactions.

Toxicity of MONPs compared with standard drugs

We predicted the toxicity level of drugs and NPs using the online tool PRO TOX-II (https://tox-new.charite.de/ protox_II/).^{37,38} This server predicted the LD_{50} (Lethal dose of 50%) of drugs and NPs using input files in mol format.

Results

Docking results of anti-bacterial targets

The binding energy in kcal/mol and inhibition constant in micromolar were calculated after molecular docking for bacterial targets with PDB IDs 1AJ0, 1QME, 1RC2, 4URO, and 5V68 (Table 1). Sulfaphenazole had the best binding energy with -8.51 kcal/mol for the Dihydropteroate synthase (1AJ0) target, as shown in Table 1. Also, Fe₂O₂ had the best MONP with -6.27 kcal/mol. For second bacterial target, penicillin-binding protein 2X (1QME), the best drug and NP, were cloxacillin, and MgO, with binding energies of -8.50 and -7.87 kcal/mol respectively (Fig. 1). The best drug and NP also were found to bind to the aquaporin Z (1RC2) protein as third bacterial target. Octyl beta-D-galactopyranoside and MgO had binding energies of 3.58 and -7.97 kcal/mol, respectively. Novobiocin is the only standard drug that has been introduced for the DNA gyrase subunit B (4URO) target. It has a binding energy of -4.73 kcal/mol and also the best NP is Fe₂O₃ with binding energy of -6.17 kcal/mol (Fig. 2). The protein involved in cell division, Ftsz (5V68), is the final bacterial target investigated in this study. The best drug is 5'-guanosinediphosphate-monothiophosphate, which has a binding energy of -5.03 kcal/mol, and the best NP is MgO, which has a binding energy of -5.27 kcal/mol (Fig. 3). Figs. 1 to 3 clearly summarizes the pattern of interaction between antibacterial targets and drugs or NPs/MONPs. The key residues of the investigated targets in interactions with MONPs and drugs are also summarized in Table 1.

Docking results of anti-cancer targets

The data in Table 2 were obtained by performing molecular docking on anticancer targets with PDB IDs 1BNA, 1EMR, 1H2N, 1RE1, 2AR9, 3HDN, 3IG7, 5CQG, 5OQW, and 8DJM using AutoDock 4.2.6 software. Gentian Violet (-12.23 kcal/mol) was the best drug in terms of binding energy for the first target, human B-DNA (1BNA). The best MNOPs docked at the first target was

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Table 1. The results of the molecular docking process of drugs and MONPs with a carbon layer cap with anti-bacterial targets

Target protein (PDB ID) (organism) (Predicted binding site)	Drugs/MONPs	Binding energy (kcal/mol)	Ki (µM) Inhibition constant
	Drugs:		
	1. Acetyl sulfisoxazole	-6.39	20.80
	2. Sulfacetamide	-5.95	43.55
	3. Sulfacytine	-6.59	14.71
	4. Sulfamerazine	-6.43	19.30
	5. Sulfameter	-6.49	17.48
	6. Sulfamethazine	-6.99	7.51
	7. Sulfamethizole	-6.59	14.80
	8. Sulfamethoxazole	-6.90	8.69
Dihydropteroate synthase (1AJ0) (E. coli)	9. Sulfanilamide	-5.61	77.20
(41.899, 1.026, 9.232)	10. Sulfaphenazole	-8.51	579.59nM
	11. Sulfisoxazole	-7.12	5.99
	NPs:		
	1. Ag_O_	-5.84	52.59
	2. CdO	-4.82	295.44
	3. CuO	-4.37	627.30
	4. Fe O	-6.27	25.37
	5. FeO	-5.54	86.71
	6 MgO	-5.87	49.88
	7 MnO	-5.42	105.97
	8.7nO	-5.36	117.61
	Druge:	-3.30	117.01
	1 6-0-Cannyloylsucrose	_1 59	133 01
	2. Clovacillin	-4.59	455.94 E 90 12 mM
		-8.50	589.1211IVI
	1 Az O	ГОГ	F1 22
Penicillin-binding protein 2X (1QME)	1. Ag ₂ O ₃	-5.85	2.04
(Streptococcus penumoniae)	2.00	-7.39	3.84
(95.173. 57.183. 53.895)	3. CUO	-5.38	114.71
()	4. Fe ₂ O ₃	-6.25	26.31
	5. FeO	-7.57	2.82
	6. MgO	-7.87	1.71
	7. MnO	-7.59	2.72
	8. ZnO	-6.07	35.59
	Drugs:		
	$1. C_{19} H_{36} O_{10} P^{-}$	-2.66	11.19mM
	2. B-2-Octylglucoside	-3.59	2.33mM
	octyl alpha-L-altropyranoside	-3.18	4.69mM
	 octyl beta-D galactopyranoside 	-3.85	1.50mM
Aquaporin Z	NPs:		
(1RC2) (<i>E. coli</i>)	1. Ag ₂ O ₃	-6.20	28.34
(79.830, 7.067, -26.468)	2. CdO	-7.24	4.93
	3. CuO	-5.19	156.46
	4. Fe ₂ O ₃	-6.23	27.16
	5. FeO	-7.52	3.06
	6. MgO	-7.97	1.43
	7. MnO	-7.48	3.30
	8. ZnO	-5.71	65.62
	Drugs:		
	1. Novobiocin	-4.73	343.28
	NPs:		
	1. Ag ₂ O ₃	-5.85	51.08
DNA gyraca subunit B (ALIPO) (Stanbylacaccus	2. CdO	-5.41	108.69
	3. CuO	-4.87	270.64
aureus) (35.214, -12.539, 8.265)	4. Fe ₂ O ₂	-6.17	30.09
	5. FeO	-5.59	80.45
	6. MgO	-5.84	52.41
	7. MnO	-5.57	82.62
	8. ZnO	-5.46	98.70
	Drugs:		
	1. 5'-Guanosine-diphosphate-monothiophosphate	-5.03	204.28
	2. Guanosine-5'-diphosphate	-2.79	8.98mM
Filementing temperature constitution where 1.7	NPs:		
Filamenting temperature-sensitive mutant Z or	1. Ag O	-5.05	199.36
Htsz (5V68)	2. CdO	-4.70	356.10
(Mycobacterium tuberculosis)	3. CuO	-4.26	749.81
(-9.737, 73.159, 18.073)	4 Fe O	-5.05	197 93
	5 FeO	-5.03	204 31
	6 MgO	-5.05	136 79
	7. MnO	-3.27	2/2 /6
	8. ZnO	-4.86	272.70
	0. 200	1 .00	212.03



Fig. 1. Docking positions for the main drugs and NPs with antibacterial receptors (Dihydropteroate synthase (1AJ0) and Penicillin-binding protein 2X (1QME)) with lowest binding energy (visualized were obtained using ADT 1.5.7 and discovery studio 2021).



Fig. 2. Docking positions for the main drugs and NPs with antibacterial receptors (Aquaporin Z (1RC2) and DNA gyrase subunit B (4URO)) with lowest binding energy (visualized were obtained using ADT 1.5.7 and discovery studio 2021).



Fig. 3. Docking positions for the main drugs and NPs with antibacterial receptor (Filamenting temperature-sensitive mutant Z or Ftsz (5V68)) with lowest binding energy (visualized were obtained using ADT 1.5.7 and discovery studio 2021).

MgO, with a binding energy of -6.41 kcal/mol. For the second target, human leukemia inhibitory factor (1EMR), doxorubicin and MgO with binding energies of -6.09 and -5.36, respectively, were found to be the best drugs and MONPs (Fig. 4). The best drug, 2-Methoxyestradiol, had a binding energy of -8.19 kcal/mol for the third anti-cancer

target, hypoxia-inducible factor I (1H2N), and the best nanoparticle, MgO, had a binding energy of -7.19 kcal/ mol. The fourth anti-cancer target, caspase3 (1RE1), was then studied. In terms of binding energy, glycyrrhizic acid (-14.26) and MgO (-6.98 kcal/mol) were obtained for the fourth target (Fig. 5). The only drug introduced for the

Table 2. The results of the molecular docking process of drugs and MONPs with anti-cancer targets

Target Protein (PDB ID) (organism) (Predicted Binding site)	Drugs/MONPs	Binding energy (kcal/mol)	Ki (μM) Inhibition constant
	Drugs:		
	1. Amsacrine	-9.40	129.55
	2. Carmustine	-5.59	80.21
	3. Chlorambucil	-7.55	2.92
	4. Cladribine	-6.79	10.61
	5. Gentian Violet	-12.23	1.08
	6. Mitomycin	-8.78	363.63
	7. Pipobroman	-7.06	6.73
	8. Streptozocin	-6.17	29.80
B DNA (1BNA)	9. Tioguanine	-5.57	82.37
(11.275, 25.052, 7.195)	Nanoparticles:		
	1. Ag ₂ O ₃	-4.89	261.46
	2. CdO	-5.46	99.37
	3. CuO	-4.51	491.84
	4. Fe ₂ O ₃	-5.17	162.85
	5. FeO	-6.10	33.74
	6. MgO	-6.41	19.86
	7. MnO	-5.86	50.98
	8. ZnO	-4.96	231.90

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Target Protein (PDB ID) (organism)	Drugs/MONPs	Binding energy	Ki (μM)
(Predicted Binding site)	Druge	(KCal/mol)	Inhibition constant
	Drugs:	6.00	24.12
	1. Doxorubicili	-0.09	620.72
	2. Fluorouracii	-4.30	039.72
		F 00	196 57
U	1. Ag ₂ O ₃	-5.09	180.57
(1EMR) (39.467 -0.948, 23.953)	2. CuO	-4.00	200.99 A7E 72
(ILIMR) (39.407, -0.948, 23.933)	4. Eq. Q	-4.53	4/5./3
	4. Fe ₂ O ₃	-5.15	172.55
	5. FeO	-5.09	117.00
	6. MgO	-5.50	209.25
	7. MIIO 8. ZnO	-5.02	208.25
	8. ZIIO	-5.00	210.79
	Diugs.	9 10	005.97
	2. Carvedilal	-8.19	995.87
		-7.29	4.52
	5. EININD-1136	-7.52	4.29
	4. FG-2210	-0.44	10.00
		-5.87	49.54
	0. PA-478	-0.05	13.30
Hypoxia-inducible factor I	7. Vadadustat	-0.52	16.76
(1H2N) (19.080, 25.679, 33.882)		6.39	25.06
	1. Ag ₂ O ₃	-0.28	25.00
	2.000	-0.00	30.18
	4. Eq. Q	-5.80	25.05
	4. Fe ₂ O ₃	-0.27	25.25
	S. FEO	-0.94	5.22
		-7.19	5.38
	7. MIIO	-0.80	9.30
	8. ZIIU	-5.//	59.25
		E E E	9E E7
	$1. C_{19} T_{13} C_{10} C_{0}$	-5.55	20.62
	$2. C_{19} \Pi_{14} N_2 O_6$	-0.39	20.02
	$3. C_{20} T_{23} N_3 U_5$	-5.08	187.70
	$4. C_{20} \Pi_{20} N_2 U_5 S$	-5.95	43.09
	$5. C_9 \Pi_{11} NO_5 S_2$	-5.12	170.01
	$5. C_{17} \Pi_{14} N_2 O_6$	-0.00	13.11
	$7. C_{20} \Pi_{20} N_2 U_9 S$	-5.95	43.40
	8. Aspirin	-4.70	325.27
	10. Glycyrrhizic acid	-0.87	25.27
	10. Giycyrmizic acid	-14.20	202.62
	12. Mothyd (20) 2 [/tart hutowcarhonyd)aminal 4 ovenentaneate	-0.91	295.05
	12. Miecryclino	-4.04	20.61
(36 935 94 291 17 839)		-0.01	33.01
(30.333, 34.231, 17.833)	14. $C_{21}\Pi_{18}\Pi_{3}O_{6}$	-0.12	921.61
		-0.29	1 92
	10. FAC-1	-7.25	4.62
		-3.87	49.05
	Nanoparticles:	-2.89	7.04
	1 Δα Ο	5 07	10 60
	1. Ag ₂ O ₃	-5.67	49.00
	2.000	-0.01	14.22 61.11
	4 Eq.O	-5.72	20 51
	<u>-</u> ₂ .0 ₃	-0.01	10.46
	6 MgO	-0.79	7 61
	7 MnO	-6.85	9.52
	8 7nO	-0.03	120.60
	0. 200	-2.22	120.00

Table 2. Continued

Target Protein (PDB ID) (organism) (Predicted Binding site)	Drugs/MONPs	Binding energy (kcal/mol)	Ki (µM)
(Frederica binding site)	Drugs:		
	1. Pamidronic acid	-6.35	22.07
	Nanoparticles:		
	1. Ag_O_	-5.30	130.17
	2. CdO	-5.41	108.36
Caspase9 (2AR9)	3. CuO	-4.97	226.43
(27.831, 44.977, 9.946)	4. Fe ₂ O ₂	-6.89	8.96
	5. FeO	-6.38	21.01
	6. MgO	-6.78	10.65
	7. MnO	-6.24	26.76
	8. ZnO	-5.64	72.81
	Drugs:		
	1. Dabrafenib	-10.23	31.83
	2. Fostamatinib	-10.43	22.54
	Nanoparticles:		
Sorum and glucocorticoid regulated	1. Ag ₂ O ₃	-6.08	34.92
kinase 1 or Sak1	2. CdO	-5.01	213.45
(3HDN) (28,185, 34,646, 71,484)	3. CuO	-5.34	121.09
	4. Fe ₂ O ₃	-6.53	16.40
	5. FeO	-5.64	73.95
	6. MgO	-6.41	19.88
	7. MnO	-5.46	98.65
	8. ZnO	-6.16	30.63
	Drugs:		
	1. Doxorubicin	-8.46	626.37
	Nanoparticles:		
	1. Ag ₂ O ₃	-5.70	66.73
Cell division protein kinase2 or CDK2	2. CdO	-6.79	10.56
(31G7) (-1.776, 31.688, 10.929)	3. CuO	-4.92	246.69
	4. Fe ₂ O ₃	-5.61	/6.83
	S. FEU	-6.96	7.93
		-7.24	4.95
	7. MIIU 8. 7n0	-0.98	7.01
		-5.65	55.54
	1 Mehendazole	-7 /9	3.23
	Nanonarticles:	-7.45	5.25
	1 Δσ Ο	-6 39	20.56
	2 CdO	-6.33	22.50
Telomerase reverse transcriptase	3 (110	-4 81	296.87
(5CQG) (-22.471, 31.524, -22.574)	4. Fe O	-6.03	38.01
	5. FeQ	-6.79	10.61
	6. MgO	-8.03	1.29
	7. MnO	-6.30	23.93
	8. ZnO	-5.28	135.93
	Drugs:		
	1. C ₂ , H ₂₀ N ₄ O ₂	-8.48	612.46
	2. Dequalinium	-7.03	7.07
	3. C ₂₄ H ₂₂ N ₂ O ₄	-9.77	69.07
	4. Terpinen-4-ol	-5.08	187.56
	Nanoparticles:		
X-linked inhibitor of apoptosis protein	1. Ag ₂ O ₃	-5.55	86.17
UI AIAP (DUUW)	2. CdO	-4.30	701.78
(10.179, 0.450, -13.000)	3. CuO	-6.70	12.24
	4. Fe ₂ O ₃	-7.40	3.77
	5. FeO	-4.52	484.08
	6. MgO	-5.06	197.00
	7. MnO	-4.35	649.31
	8. ZnO	-6.81	10.11

Table 2. Continued

Target Protein (PDB ID) (organism) (Predicted Binding site)	Drugs/MONPs	Binding energy (kcal/mol)	Ki (μM) Inhibition constant
	Drugs:		
	1. Atorvastatin	-7.42	3.63
	2. Cerivastatin	-7.71	2.24
	3. Fluvastatin	-8.58	511.79
	4. Lovastatin	-5.95	43.19
	5. Mevastatin	-6.00	40.33
	6. Pitavastatin	-7.58	2.76
	7. Pravastatin	-4.74	334.93
	8. Rosuvastatin	-6.68	12.64
HMG COA Reductase (8DJM)	9. Simvastatin	-6.27	25.32
(145./24, 111.5/1, 150.161)	Nanoparticles:		
	1. Ag ₂ O ₃	-5.83	53.62
	2. CdO	-5.87	49.69
	3. CuO	-4.68	372.13
	4. Fe ₂ O ₃	-5.62	75.98
	5. FeO	-6.11	33.12
	6. MgO	-6.46	18.39
	7. MnO	-6.04	37.61
	8. ZnO	-5.59	79.72

fifth target, caspase9 (2AR9), had a binding energy of -6.35 kcal/mol, and the best MONP was also Fe₂O₃ with a binding energy of -6.89. The best drug for the sixth anticancer target, serum and glucocorticoid-regulated kinase 1 (3HDN), was Fostamatinib, with a binding energy of -10.43 kcal/mol, and the best MONP was Fe₂O₂, with a binding energy of -6.53 (Fig. 6). Following the molecular docking process, other drugs and nanoparticles with other anticancer targets were investigated. The only introduced drug, doxorubicin, had a binding energy of -8.46 kcal/ mol for the seventh target, cell division protein kinase 2 (3IG7), and the best MONP was identified as MgO with a binding energy of -7.24 kcal/mol. The eighth target is also its only drug, mebendazole, which has a binding energy of -7.49 kcal/mol, and the best MONP is MgO, which has a binding energy of -8.03 kcal/mol (Fig. 7). Regarding the ninth target (XIAP), with PDB ID 50QW, the best drug is N-Methylalanyl-3-methylvalyl-4-phenoxy-N-(1,2,3,4tetrahydronaphthalen-1-YL)prolinamide $(C_{21}H_{42}N_4O_4)$ with a binding energy of -9.77 kcal/mol, and the best MONP is Fe₂O₃ with a binding energy of -7.40 kcal/ mol. The best drug and MONP for the 10th anticancer target, the HMG CoA reductase enzyme (8DJM), were Fluvastatin and MgO, with binding energies of -8.58 and -6.46 kcal/mol, respectively (Fig. 8). Table 2 shows complementary results of how the best drugs and MONPs bind. Figs. 4 to 8 clearly shows the key residues that play a role in the interactions of targets with drugs or MONPs in a summary form and with a specified pattern.

The efficiency of standard drugs and carbon capped-MONPs

The antibacterial properties of the five targets under study

were investigated in this study, and in the case of two targets, 1AJ0 and 1QME, the target's standard drug had the highest binding energy and the lowest inhibition constant and performed better than MONPs, but other targets, namely 1RC2, 4URO, 5V68, MgO, and Fe₂O₃ MONPs, performed better. By targeting Aquaporin Z, DNA gyrase subunit B, and FtsZ, these two nanoparticles showed good antibacterial properties. Also, when compared to standard drugs, MONPs have different binding to target proteins, as shown in Table 2, only two MONPs of Fe₂O₃ and MgO, outperformed conventional drugs in the case of the anti-cancer targets, out of the 10 investigated targets. Both the telomerase reverse transcriptase (5CQG) and caspase-9 (2AR9) were more effectively targeted by Fe₂O₂ NP and MgO NP, respectively. However, in the eight other anticancer targets studied, the standard drugs performed better than MONPs. As illustrated in Figs. 9 A and B, the binding energy distribution of MONPs for all the targets investigated in this study was between -4 and -8 kcal/mol, whereas the distribution of drugs is between -2.5 and -14.5 kcal/mol. MONPs have shown a minor difference in binding energy for different targets in these diseases, which is likely because of the unique physicochemical and pharmacophoric properties of MONPs.^{39,40} These properties have resulted in NPs performing similarly to various targets.

The toxicity of standard drugs and carbon-capped MONPs

MONPs were more toxic than other drugs in comparison studies between NPs and standard drugs, as shown in Table 3 and Fig. 10 (LD_{50} , or lethal dose 50%, is the concentration of a compound that causes the death of



Fig. 4. Images of the interactions involved between the best drugs and MONPs with anti-cancer targets (B DNA (1BNA) and Human-leukemia inhibitory factor (1EMR)) based on molecular docking results.



Fig. 5. Images of the interactions involved between the best drugs and MONPs with anti-cancer targets (Hypoxia-inducible factor I (1H2N) and Caspase 3 (1RE1)) based on molecular docking results.



Fig. 6. Images of the interactions involved between the best drugs and MONPs with anti-cancer targets (Caspase9 (2AR9) and Serum and glucocorticoidregulated kinase 1 or Sgk1(3HDN)) based on molecular docking results.



Fig. 7. Images of the interactions involved between the best drugs and MONPs with anti-cancer targets (Cell division protein kinase2 or CDK2 (3IG7) and Telomerase reverse transcriptase (5CQG)) based on molecular docking results.



Fig. 8. Images of the interactions involved between the best drugs and MONPs with anti-cancer targets (Cell division protein kinase2 or CDK2 (3IG7) and Telomerase reverse transcriptase (5CQG)) based on molecular docking results.



Fig. 9. Comparison of anti-Cancer (A) anti-Bacterial (B) molecular docking results

50% of test samples, especially in the case of cancer cells and bacteria). Another issue is that MONPs have a high molecular weight when compared to standard drugs. Because of their high molecular weight, these NPs are likely to be toxic to various organs of the human body if taken orally. However, such problems can be avoided by adjusting the dosage and application of biocompatible natural materials such as herbal metabolites so that the lowest one has a level of toxicity for the organs of the human body.⁴¹⁻⁴³

Discussion

Traditional drug discovery depending on *in vitro* and *in vivo* studies has several disadvantages of time-consuming processes, large investments for experimental set-ups,

Table 3	. The	predicted	toxicity	of MONPs	by P	RO TOX-II
Tuble 0		predicted	toxicity		Dyi	

MONPs	LD ₅₀ (mg/kg)	MW (g/mol)	
Ag ₂ O ₃	900	1109	
CdO	200	2617	
CuO	2000	873.92	
Fe ₂ O ₃	68	879.44	
FeO	305	1774.47	
MgO	1000	1311.75	
MnO	68	1637.8	
ZnO	1000	993.73	

MONPs: Metal oxide NPs; LD₅₀: Lethal dose 50%; MW: Molecular weight

and feasibility only to biopharmaceutical companies. In this case, to save time and resources, screening of repurposable chemical materials relying on modern artificial intelligence-based algorithms and state-ofthe-art computational techniques can be applied to discover new drugs.44 Rapid and cost-effective in silico methods based on structure-based drug repurposing have a critical role in targeting specific receptors related to health-threatening diseases such as cancers, and diabetes by natural compounds and NPs.45 Drug/target network-based models, modern artificial intelligence, and structure-based approaches have been used for the viral pandemic. However, there are some limitations to these computational drug repositioning/repurposing/ reprofiling approaches. For example, the results of modern artificial intelligence-based networks are less relevant due to the necessity of more data for their application.⁴⁶ In addition, a low chemical library can lead to the limited efficiency of molecular docking approaches.⁴⁷ Therefore, in vitro and in vivo studies should be considered to overcome these disadvantages.48

Based on the toxicity property and efficiency of the MONPs tested in this study, it can be concluded that biocompatible MONPs have therapeutic potential for targeting specific receptors. However, the role of these NPs in ROS production should not be underestimated, because previous studies found that many of the antibacterial and anticancer properties of NPs were due to the production of this active species of free radicals.⁴⁹ For example,



Fig. 10. Comparison of the toxicities of MONPs and conventional medications with anticancer (A) and antibacterial (B) activities is shown in two graphs. MONPs, as is well known, have a lower LD50 and a higher molecular weight than standard drugs, and NPs exhibit greater toxicity.

MgO NPs caused apoptosis in K-562 lymphoblast cell lines by the production of ROS. Moreover, molecular docking results for this study showed that MgO NPs with a diameter of 1.5 nm can interact with human serum albumin via hydrophobic residues using hydrophobic forces.⁵⁰ In a comparative study, docking of four nanocomposites including Fe₂O₄@SiO₂, Fe₂O₄@SiO₂@ APTS (3-aminopropyl) triethoxysilane), Fe₂O₄@SiO₂@ APTS~Schiff base, and Fe₂O₄@SiO₂@APTS~Schiff base-Cu(II) were evaluated against DNA duplex of sequence d (ACCGACGTCGGT), ribonucleotide reductase, and topoisomerase II. Highest binding affinity was found for docking of Fe₃O₄@SiO₂@APTS with DNA and Fe₃O₄@ SiO₂@APTS~Schiff base-Cu(II) with topoisomerase II by values of -10.85 and -7.94 kcal/mol, respectively.⁵¹ In another study, the best binding energy value between AgO NPs and Als3 adhesin was -9.60 kcal/mol. In addition, these NPs exhibited an IC₅₀ value of 36.56 μ g/mL against A431 epidermoid carcinoma cell line.⁵² One of the most important aspects of using nanoparticles as therapeutic agents is their ease of penetration into the structures of proteins or other targets.53 It is worth noting that the toxicity of drug agents is not limited to cancer cells; these agents must also be able to respond appropriately to the tumour microenvironment.54 Introducing MONPs as potential agents in the field of cancer therapy can be effective if the primary characteristic of the disease, namely its complexity, is considered. The agents may have anticancer properties against cancer cells in vitro, but they are

Research Highlights

What is the current knowledge?

 $\sqrt{}$ One of the main challenges in the development of drugs based on MONPs is their toxicity based on LD_{so}.

What is new here?

 $\sqrt{}$ The toxicity of MONPs when used as an oral drug was higher than standard drugs, although their toxicity needs further investigation in the future.

ineffective *in vivo*.⁵⁵ Due to their unique physicochemical properties, nanoparticles have the potential to be among the leading candidates in cancer chemotherapy.

Conclusion

The results of this study indicate that MONPs have therapeutic potential for use as oral medications against a variety of disease-related targets. These findings undoubtedly confirm the anticancer and antibacterial properties reported in earlier studies. One of the major challenges in the development of drugs based on MONPs is their toxicity as measured by LD₅₀, so this important parameter was investigated in this study. The toxicity of MONPs as oral drugs was higher than standard drugs, but this needs to be investigated further in the future. MgO and Fe₂O₂ NPs have the highest efficiency among the MONPs under investigation, according to the data obtained from these molecular docking studies. One intriguing finding was that these two MONPs had a greater ability to bind to specific targets for cancer, and bacterial infection than other MONPs. It is even greater than the standard drugs for these targets. The toxicity level of MgO (1000 mg/kg) was determined to be much lower than that of Fe₂O₃ (68 mg/kg). In this context, it is clear that bioinformatics tools for drug development based on MONPs must be developed. In addition, the best MONPs must be developed and designed to have higher efficacy and lower toxicity.

Authors' Contribution

Conceptualization: Navid Mohammadjani, Morahem Ashengroph, and Mehran Alavi.

Data curation: Navid Mohammadjani, Mehran Alavi, Morahem Ashengroph, and Sahand Karimi.

Investigation: Navid Mohammadjani, Morahem Ashengroph, Sahand Karimi, Mehran Alavi, and Musa Moetasam Zorab.

Project administration: Morahem Ashengroph and Mehran Alavi.

Supervision: Morahem Ashengroph and Mehran Alavi. **Validation:** Morahem Ashengroph, Mehran

Validation: Morahem Ashengroph, Mehran Alavi, Navid Mohammadjani, Sahand Karimi, and Musa Moetasam Zorab.

Visualization: Navid Mohammadjani, Morahem Ashengroph, Mehran Alavi, Sahand Karimi, and Musa Moetasam Zorab.

Writing-original draft: Navid Mohammadjani, Sahand Karimi, and Musa Moetasam Zorab.

Writing-review & editing: Mehran Alavi, Navid Mohammadjani, and Morahem Ashengroph.

Competing Interests

No potential conflict of interest was reported by the authors.

Ethical Statement

None to be declared.

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