







In vitro study of the anticancer activity of various doxorubicincontaining dispersions

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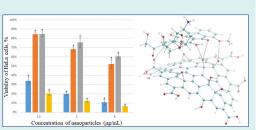
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Abstract

Introduction: The aim of this research was to study the impact of various doxorubicin (Dox)-containing nanofluids, e.g. single-walled carbon nanotube (SWCNT)+Dox, graphene oxide (GO)+Dox and Dextran-PNIPAM (copolymer)+Dox mixtures on HeLa cells (human transformed cervix epithelial cells, as a model for cancer cells) depending on their concentration.



Methods: Structural analysis of GO+Dox complex was accomplished using Hartree-Fock level of theory in 6-31G** basis set in Gaussian. Dynamic light scattering (DLS), zeta-potential, scanning electron microscopy and confocal laser scanning microscopy were used. The cell viability was analyzed by the MTT assay.

Results: The viability of HeLa cells was studied with the MTT assay after the incubation with various Dox-containing dispersions depending on their concentration. The size of the particles was determined by DLS. The morphology of the nanoparticles (NPs) was studied by scanning electron microscopy and their uptake into cells was visualized by confocal laser scanning microscopy. It was found that the Dextran-PNIPAM+Dox nanofluid in contrast to Dox alone showed higher toxicity towards HeLa cells up to 80% after 24 hours of incubation, whereas the SWCNT+Dox and GO+Dox nanofluids at the same concentrations protected cells from Dox.

Conclusion: The importance of Dextran-PNIPAM copolymer as a universal platform for drug delivery was established, and the huge potential of Dextran-PNIPAM+Dox NPs as novel anticancer agents was noted. Based on the in vitro study of the SWCNT+Dox and GO+Dox nanofluids, it was concluded that SWCNT and GO NPs can be effective cytoprotectors against the highly toxic drugs.

Introduction

Nanoparticles (NPs) have the advantage of targeting cancer by simply being effectively accumulated and entrapped in tumors due to the phenomenon of enhanced permeability and retention caused by leaky tumor angiogenetic vessels and poor lymphatic drainage.^{1,2} This effect constitutes one of the major advantages of NPs against multiple drug resistance (MDR) mechanisms.³ Many of the drugs available to circumvent MDR were in the past unavailable to target this mechanism due to low solubility and/or stability. NPs made these drugs a possible strategy to target MDR. They have also provided an effective platform to deliver high loads of drugs in a specific and controlled way (using stimuli-sensitive and pH-responsive), with surface modified to improve circulation time, to prevent the uptake by the reticuloendothelial system and to ensure specific targeting (using bioactive agents like antibodies, receptor ligands, siRNA). Another advantage of using NPs for MDR regression is the drastic reduction of the IC_{50} value for drugs thus reducing the clinical doses. In

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fact, the lipid nanocapsules, polymeric micelles, metal and carbon NPs have been reported to circumvent drug resistance.^{4,5} In this regard, the conjugation of anticancer drugs, for example, doxorubicin (Dox) as a 'gold standard' of chemotherapy, with the representatives of carbon NPs could be promising for targeted drug delivery, overcoming of tumor cells drug resistance and reduction of side toxic effects.⁶⁻⁸

In the present work, we studied the impact of various Dox-containing nanofluids, *e.g.* single-walled carbon nanotube (SWCNT)+Dox, graphene oxide (GO)+Dox and Dextran-PNIPAM (copolymer)+Dox mixtures on HeLa cells (human transformed cervix epithelial cells, as a model for cancer cells) depending on their concentration.

Materials and Methods Material preparation

Dox (Doxorubicin hydrochloride; Sigma-Aldrich, USA) was dissolved in saline with a maximum concentration of 0.15 mg mL^{-1} .

An aqueous solution of carboxylated SWCNT (Sigma-Aldrich, USA) in a maximum concentration of 0.15 mg mL⁻¹ was prepared as described in previously published papers.^{9,10}

GO (Sigma-Aldrich, USA) was dispersed in ultrapure water to prepare a stock solution with a maximum concentration of 0.15 mg mL⁻¹. Then this aqueous solution was sonicated for 2 hours (40 kHz).¹¹

Preparation of the SWCNT+Dox and GO+Dox mixtures were executed according to the protocol: SWCNT or GO and Dox were mixed in a 1:1 volume ratio (0.075:0.075 mg), and the resulting aqueous solution was treated in the ultrasonic disperser for 15 minutes, and after that stirred magnetically for 24 hours at room temperature.

The Dextran-PNIPAM copolymer used as a nanocarrier for Dox has been described in a recently published paper.¹² This copolymer is star-like with Dextran core and PNIPAM grafts (M_w=1.03×10⁶ g mol⁻¹; M_w/ $M_{p}=1.52$; $D_{p}=38$ nm). Dextran-PNIPAM exhibits a lower critical solution temperature (LCST). The polymer is soluble in water below LCST and undergoes a phase transition above LCST becoming partially hydrophobic. The conformational transition for this copolymer was registered in the temperature range of 32.6-33.4°C, that higher than typical LCST point for linear PNIPAM of similar molecular weight and polydispersity and is closer to physiological temperature (37°C).¹³ A possible tuning of the hydrophobicity of star-like polymer, the regulation of the region of phase transition and size of hydrophobic domains by variation of copolymer internal structure was reported.12

A stock solution of Dextran-PNIPAM (0.15 mg mL⁻¹) was prepared in distilled water. This concentration is lower of a crossover concentration.¹³ For obtaining an aqueous solution of Dextran-PNIPAM+Dox, the Dextran-PNIPAM and Dox were mixed in a 1:1 volume

ratio (0.075:0.075 mg).

Structural calculations

The structural parameters of GO molecule were taken from¹⁴ and the initial structure was built with an aid of HyperChem8 software. After that, the structure of Dox molecule, taken from Protein Data Bank (PDB ID 1D12), was added in HyperChem. The planes of GO sheet and Dox chromophore were located parallel to each other at a distance of 3.4 Å which is typical of aromatic-aromatic stacking. The longitudinal axis of the Dox molecule was located above the central row of carbon cycles of the GO sheet. Preliminary optimization of this structure by means of molecular mechanics was performed in HyperChem using AMBER force field. After that full minimization by the energy of the spatial structure of GO+Dox complex was accomplished using Hartree-Fock level of theory in 6-31G** basis set in Gaussian.

Dynamic light scattering and zeta potential determination

Measurements of size distribution and zeta potential for various aqueous mixtures containing different particles were performed by dynamic light scattering (DLS) on a Zetasizer Nano-ZS90 (Malvern, Worcestershire, UK) at T=298 K. The instrument was equipped with a He-Ne laser (5 mW) operating at the wavelength of 633 nm. The autocorrelation function of the scattered light intensity was analyzed by the Malvern Zetasizer software with the Smoluchowski approximation.

Scanning electron microscopy

NPs presented in the studied samples were visualized by scanning electron microscopy (SEM) with an ESEM Quanta 400 instrument (FEI, The Netherlands) on gold/ palladium-sputtered samples.

Cell culture

HeLa cells were cultured in Dulbecco's modified Eagle medium (DMEM), supplemented with 10% fetal calf serum (FCS), 100 U mL⁻¹ penicillin, and 100 mg mL⁻¹ streptomycin at 37°C in humidified atmosphere with 5% CO_2 . 12 hours prior to the uptake experiments, the cells were trypsinized and seeded in 24-well plates with 5·10⁴ cells per well in 0.5 mL DMEM with FCS for HeLa cells.

The cell line was obtained from the Collection of Microorganisms and Cell Cultures of the University of Duisburg-Essen.

Confocal laser scanning microscopy

The uptake studies of NPs presented in the samples were carried out as follows: $30 \ \mu\text{L}$ of the nanofluid was added to the cells. The cells were incubated with the NPs for 3 and 48 hours, respectively. Immediately after the indicated time, the cells were washed three times with phosphate-buffered saline (PBS) in order to remove all dispersed NPs,

fixed with 4% paraformaldehyde at room temperature for 10 minutes, then stained with fluorescent dye DAPI for the cell nucleus and washed three times with PBS. Finally, the cells were studied by confocal laser scanning microscopy (CLSM; analysis of z-stacks), which gave us information whether the NPs were able to enter the cells. CLSM was performed on a TCS SP8 AOBS system controlled by the LASX software (Leica Microsystems). The laser lines were used for the excitation by DPSS 561 nm (for Dox) and Diode 405 nm (DAPI). Images were acquired with a 20x objective.

MTT assay

The cell viability was analyzed by the MTT assay 24 and 48 hours after the incubation of cells with the NPs, so the final concentration of NPs per well after dilution in the medium was: 1.5, 3 and 6 µg mL⁻¹. MTT (3-(4.5-dimethylthiazol-2-yl)-2.5-diphenyl-tetrazolium bromide; Sigma, Taufkirchen, Germany) was dissolved in PBS (5 mg/mL) and then diluted to 1 mg mL⁻¹ in the cell culture medium. The cell culture medium of the incubated cells was replaced by 300 μ L of the MTT solution. The cells were then incubated for 1 hour at 37°C under 5% CO, in a humidified atmosphere. About 300 µL of dimethyl sulfoxide (DMSO) was added to the cells. After 30 min, a 100 µL aliquot was taken for spectrophotometric analysis with a Multiscan FC instrument (Thermo Fisher Scientific, Vantaa, Finland) at λ =570 nm. The absorption of incubated cells was normalized to that of control (untreated) cells, thereby indicating the relative level of cell viability.

Statistics

Statistical analysis was performed by conventional methods of variation statistics. The significance of the differences between the control and experimental measurements was estimated within the framework of the Student's *t*-test using Origin 8.0 software (OriginLab Corporation, USA). The difference between the compared values was considered to be significant at P < 0.05.

Results and Discussion

Material characterization

The freshly prepared aqueous mixtures were characterized by DLS and SEM, in order to determine the aggregation state of NPs. Such information is considered important as the particles size is known to correlate with their bioactivity.¹⁵⁻¹⁹

Table 1 shows the experimental hydrodynamic sizes of various particles in aqueous solutions at 25°C. The Dextran-PNIPAM+Dox and SWCNT+Dox systems contain NPs with a hydrodynamic size of 69 to 81 nm. The GO+Dox mixture formed particles with hydrodynamic size 158 nm. The polydispersity index (PDI) falls in between 0.3-0.5 for all of the studied samples and indicates the degree of particle's aggregation.

Table 1. The hydrody	namic diameter of particle	s (Z-average size),
polydispersity index (F	PDI) and zeta potential fo	or various aqueous
systems at T=25°C		

Sample	Size (nm)	PDI	Zeta potential (mV)
Dextran-PNIPAM+Dox	69	0.36	-9
SWCNT+Dox	81	0.49	-13
GO+Dox	158	0.39	-10

The zeta potential study has shown that at 25°C for the Dextran-PNIPAM+Dox, GO+Dox and SWCNT+Dox aqueous systems it takes values in between -9 and -13 mV (see Table 1). A negative charge of the individual particles plays a significant role in the stabilization of the studied aqueous systems. It contributes to the electrostatic repulsion between the negatively charged clusters, thus disfavoring the aggregation and making the solution colloidally stable. For all studied systems, no signs of sediment and changes in UV-Vis absorption spectra were noted within the period of the experiment, suggesting the stability of the investigated mixtures.

Fig. 1 shows the SEM data for the studied samples, which provides additional confirmation of their polydispersity.

Structural analysis of the GO+Dox nanofluid was performed by means of a molecular modeling approach. It is known that both, GO and SWCNT, bind small aromatic molecules by formation of π -stacks with them in solution.^{6,20,21} The structure of the SWCNT+Dox complex is already known⁶ and was not investigated in this work. Hence, further analysis was focused on GO+Dox system.

It was initially assumed that the GO+Dox nanofluid contain GO+Dox complexes, which were studied by theoretical calculation of their spatial structures minimized by energy. Fig.2B contains the resultant structure of 1:1 complex. As a consequence of the structure optimization,

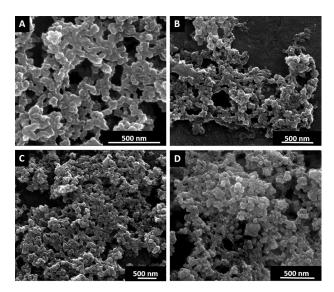


Fig. 1. Scanning electron microscopy images of Dox (A), SWCNT+Dox (B), GO+Dox (C) and Dextran-PNIPAM+Dox (D).

the initially flat sheet of GO has become bent accepting the shape of groove with the bending line passing over the central row of the carbon cycles (see dashed line in Fig. 2A). Such bending of GO sheets is well known (e.g.).²² The optimized structure features the average distance between the carbon atoms of the Dox chromophore equal to 4.3 Å. The longitudinal axis of the Dox chromophore has shifted and located above the neighboring row of the carbon cycles (shown in Fig. 2A by the solid line). In general, as a consequence of energy minimization the stacking of GO and Dox molecules within the 1:1 complex has been preserved, although the resultant distance has appeared to be relatively high (~4.5 Å).23 In addition, the obtained geometry of the studied complex allows the formation of 4 intermolecular hydrogen bonds between the GO and Dox molecules (see Fig. 2B), which, obviously, make a significant contribution to its stabilization. The possibility of intermolecular H-bonding within the π -stacked complexes of aromatic molecules including Dox is well known,²⁴ and therefore, may be expected in GO+Dox nanofluid as well.

In vitro experiment

As can be seen from the Fig. 3A, Dox already kills about 65% of cells for 24 h at the lowest used concentration of 1.5 mg mL⁻¹, and Dextran-PNIPAM+Dox almost 80% at the statistically significant level of difference. The effectiveness of these therapeutic drugs only increases

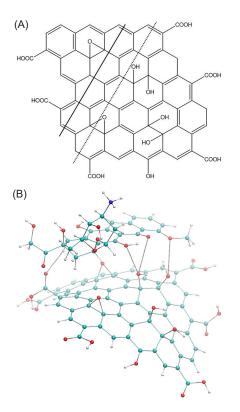


Fig. 2. The structure of GO molecule (A) and the minimized structure of GO+Dox complex (B)

with increasing concentration (Fig. 3A, B). As one can see, the Dextran-PNIPAM+Dox nanofluid shows a much higher toxicity towards HeLa cells in comparing with Dox alone.

At the same time, the SWCNT+Dox and GO+Dox nanofluids at the concentrations used are of low toxicity to cells, especially at the lowest concentration 1.5 mg mL⁻¹ (Fig. 3A, B), and also with increasing time of incubation time with the cells (Fig. 3B). Thus, it can be stated that the SWCNT and GO NPs protect cells from Dox. This observation can be explained using the following rationale. The results of the structural analysis performed above, as well as the literature data (e.g.^{6, 20, 21}), suggest the complex formation of the SWCNT and GO NPs with Dox. We assume that this effect reduces the concentration of free Dox molecules able to induce cell death, thus leading to cell protection from the antibiotic action when Dox is administered together with NPs, as compared with the Dox action as a single agent. The mechanism described is well known in literature with respect to small molecules, called as 'interceptors' or 'scavengers', which can bind with Dox via π -stacking (see ^{24,25} for details), and has recently been revealed for Dox binding with another carbon nanoparticle, i.e. C60 fullerene.7,8 Most likely, the molecular complexations act here as the universal physicochemical factor, which is independent of the type of biosystem under study.

Finally, the Dox, Dextran-PNIPAM+Dox, SWCNT+Dox and GO+Dox NPs demonstrated even more pronounced treatment and protective effects (about 80-90% cells in

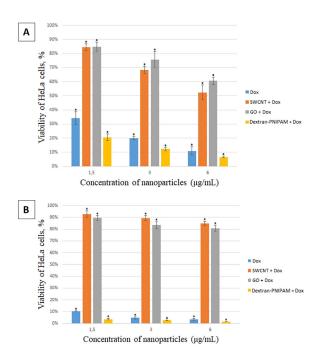


Fig. 3. MTT assay on HeLa cells after incubation with studied NPs at different concentrations for 24 hours (A) and 48 hours (B). *significant differences compared to the control (untreated cells) at P < 0.05.

both cases), respectively, when incubated with the cells for 48 hours (Fig. 3B).

It should be noted that SWCNT, GO and Dextran-PNIPAM NPs are non-toxic *in vitro* and *in vivo* in the concentration range under study (up to 6 mg mL⁻¹).²⁶⁻²⁸

It is known that Dox-induced toxicity and resistance are major obstacles in chemotherapeutic approaches. The mechanisms of toxicity and resistance are respectively related to induction of reactive oxygen species and upregulation of ATP-binding cassette transporter.²⁹ Thus, one can assume that coadministration of SWCNT and GO NPs with a high Dox dose is capable of ameliorating its cytotoxicity through their antioxidant potential.^{15,26,28,30,31}

It is important to emphasize that the Dextran-PNIPAM copolymer is thermosensitive with a transition temperature from the hydrophilic to the hydrophobic state of the macromolecule in the region of physiological temperatures.12 Therefore, one can assume that the anticancer activity of Dextran-PNIPAM+Dox nanofluid sharply increases due to a change in the conformation of the Dextran-PNIPAM copolymer, which promotes the release of Dox molecules. Thus, the novel Dextran-PNIPAM copolymer can be used as a potential carrier of anticancer drugs (for example, Dox) as opposed to SWCNT and GO NPs. As shown above, both GO and SWCNT act as interceptors of Dox molecules due to the formation of GO+Dox and SWCNT+Dox complexes, thus reducing the concentration of monomeric Dox molecules able to induce cell death.

The in vitro uptake of NPs by HeLa cells was studied by CLSM. For this purpose, we used Dox as a fluorescing marker (~500 nm; Dox gives a red fluorescence).^{31,32} Fig. 4 shows the cellular uptake and the distribution of SWCNT+Dox (A) and GO+Dox (B) after 48 h of incubation on HeLa cells. NPs were thoroughly washed in all these cases to remove adhering or dispersed NPs. Fig.

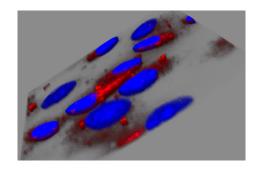


Fig. 5. Analysis of z-stacks carried out by laser confocal scanning microscopy on HeLa cells after incubation with GO+Dox (B; 48 h; $6 \ \mu g \ mL^{-1}$) NPs. This information could prove, that NPs were able to go inside a cell.

5 demonstrates the z-stack through the cells, which were carried out to prove the accumulation of GO+Dox inside the cells and not on their surface. It is important to note that in the case of Dextran-PNIPAM+Dox NPs, the results for CLSM almost are no different from those we obtained earlier for pure Dox at 3 hours of incubation with HeLa cells.³³ This again confirms the fact that the water-soluble Dextran-PNIPAM copolymer acts as the Dox molecules transporter into cells, actively releasing them to the intracellular space at physiological temperature.¹² Thus, one may conclude that the Dextran-PNIPAM+Dox NPs can be used for further pre-clinical trials.

Conclusion

This study clearly demonstrates that water-soluble Dextran-PNIPAM+Dox NPs sharply decreased the viability of HeLa cells at the low concentrations (1.5-6 μ g mL⁻¹) for 24-48 hours. This result indicates on the importance of Dextran-PNIPAM copolymer as a universal platform for drug delivery, and, in particular, the huge potential of Dextran-PNIPAM+Dox NPs as novel anticancer agents.

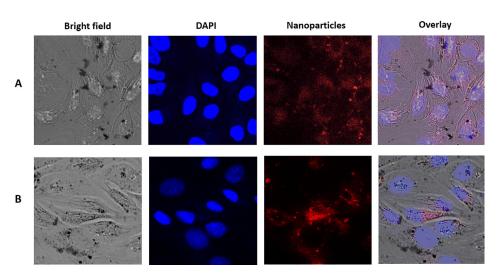


Fig. 4. Laser scanning confocal microscopy images of HeLa cells after incubation with SWCNT+Dox (A, time of incubation 48 h; concentration 6 μg mL⁻¹) and GO+Dox (B; 48 h; 6 μg mL⁻¹) NPs.

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Research Highlights

What is the current knowledge?

 $\sqrt{$ Importance of Dextran-PNIPAM copolymer as a universal platform for drug delivery is established.

What is new here?

 $\sqrt{}$ Water-soluble Dextran-PNIPAM+Dox NPs sharply decrease the viability of HeLa cells at low concentration; $\sqrt{}$ GO particles can be effective cytoprotectors against the high toxic drugs.

At the same time, the SWCNT+Dox and GO+Dox nanofluids are of low toxicity to HeLa cells at the same conditions, and hence, the SWCNTs and GO NPs can be effective cytoprotectors against the high toxic drugs. Moreover, the SWCNTs, as well as GO NPs, offer a great potential for upcoming nanomedicines nanoscaled devices, including biosensor and bioimaging techniques.

Finally, the uptake of the NPs into cancer cells was shown by CLSM.

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Ethical approval

None to be declared.

Competing interests

Authors declare no conflict of interest.

Authors contribution

The work presented here was carried out in collaboration between all the authors. SP, YH, NK and UR synthesized and characterized nanomaterials. VS and TM performed *in vitro* and microscopy studies. VK and ME performed the computer simulation. M. Epple and YP coordinated the experimental work, analyzed the data, performed the statistical analysis, and wrote the manuscript. All authors discussed the results and commented on the manuscript. All authors read and approved the final manuscript.

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